

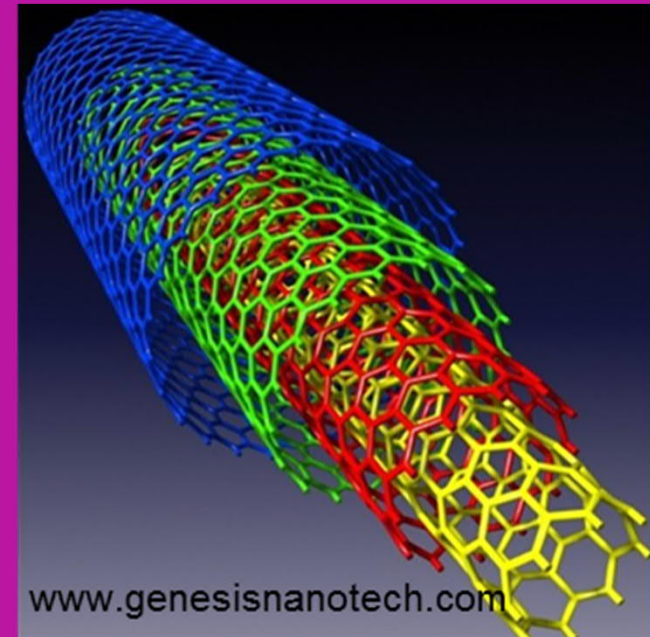
# Module C1: Durability - Nanostructures

AAE-E3120 Circular  
Economy for Energy Storage

Prof. Annukka Santasalo-Aarnio



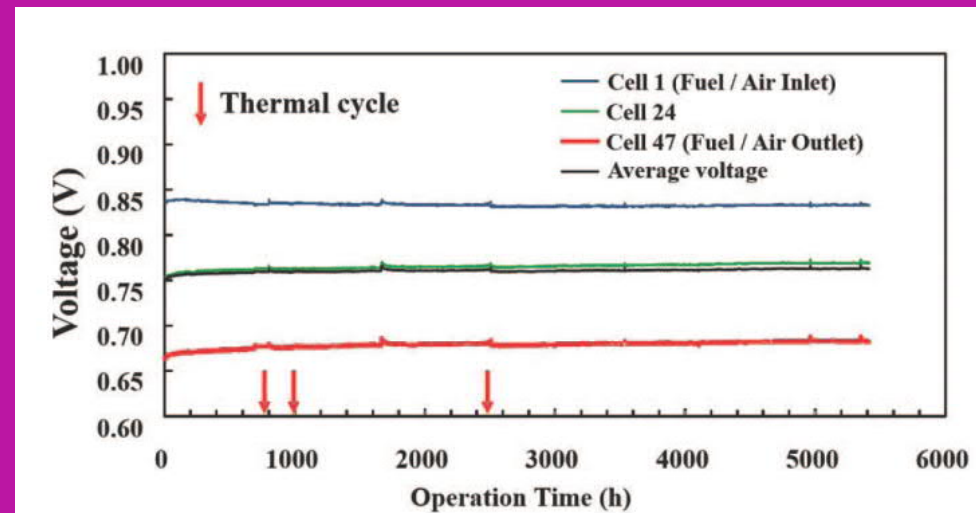
Aalto University  
School of Engineering



# Learning outcomes

- **Recognize the material choice effect to degradation mechanisms of the system**
  - Degradation mechanism in
  - Active materials (nanomaterials)
- **Develop new design for recycling approach for energy storage application and justify with scientific argumentation**
  - High durability (how to ensure with material selection?)

# Definition of Durability



## Solid Oxide Fuel Cell (SOFC)

Y. Kobayashi et al. Journal of The Electrochemical Society, 161 (2014) F214

# Durability ?

## Lecture Journal

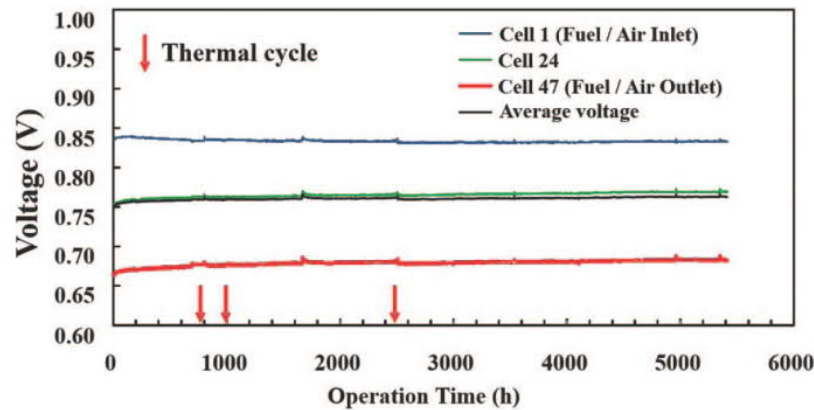
How to define durability?

How often you are willing to

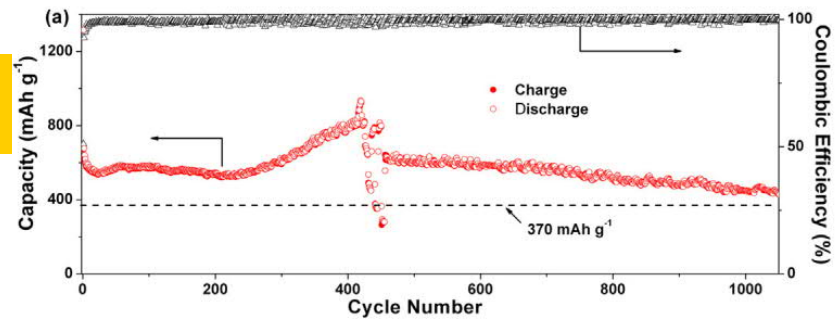
- Charge a mobile phone?
- Change the solar panel of your summer cottage?
- Change the fuel cell modules of your car?

# Durability ?

## Close to market applications



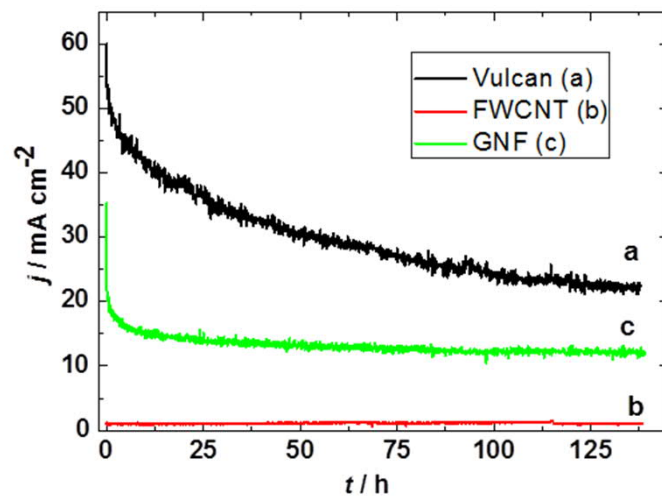
**Solid Oxide Fuel Cell (SOFC)**  
Y. Kobayashi et al. Journal of The Electrochemical Society, 161 (2014) F214



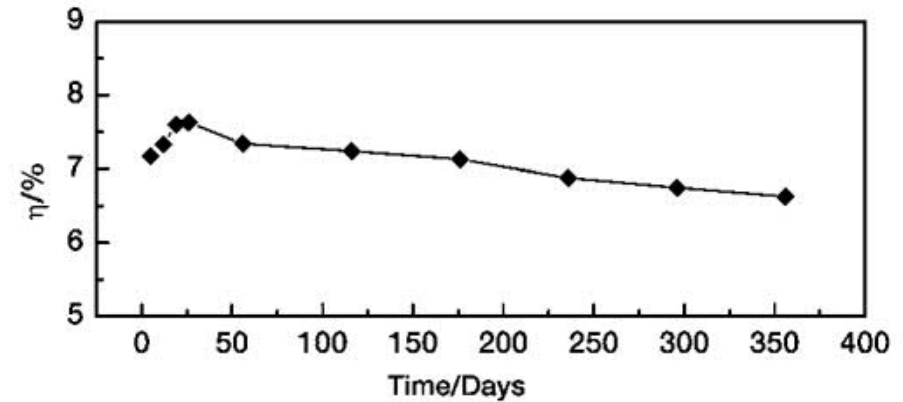
**Li-ion batteries**  
X. Zhang et al. / Journal of Power Sources 268 (2014) 365-371

# Durability ?

## Early stage applications



Direct Methanol Fuel Cell 70 °C. A. Santasalo-Aarnio et al. Int. J. Hydr. Ene. 37 (2012) 3415.

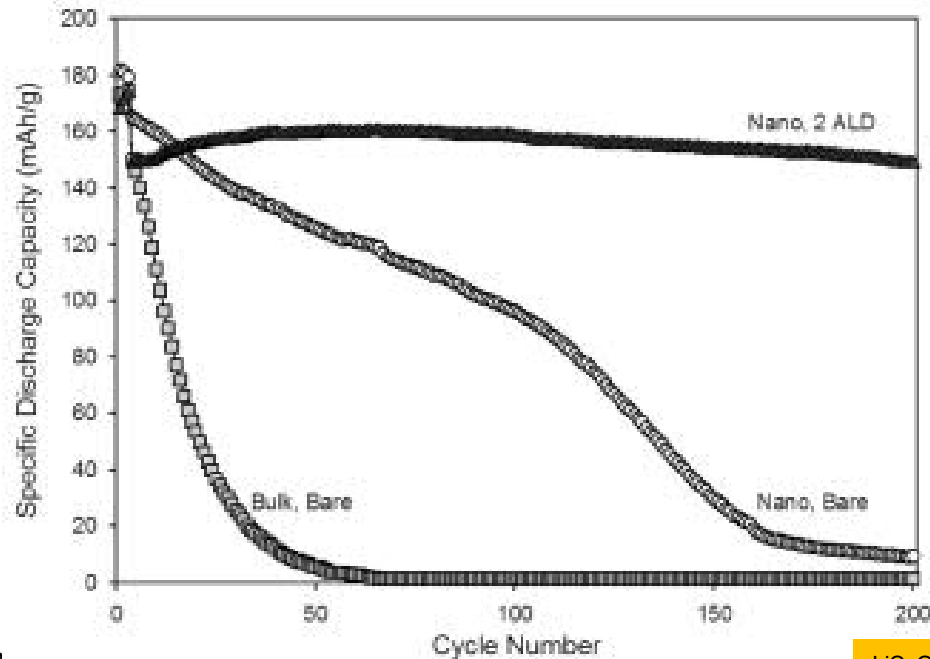


Dye-sensitized solar cell, TiC/Pt counter electrode, Wu et al. J.Mater.Chem.A (2013), 1, 9672.

# Material degradation

- How much is accepted?
- When does the application become useless?

What is presented as reference data?



Do you trust this Bulk data?

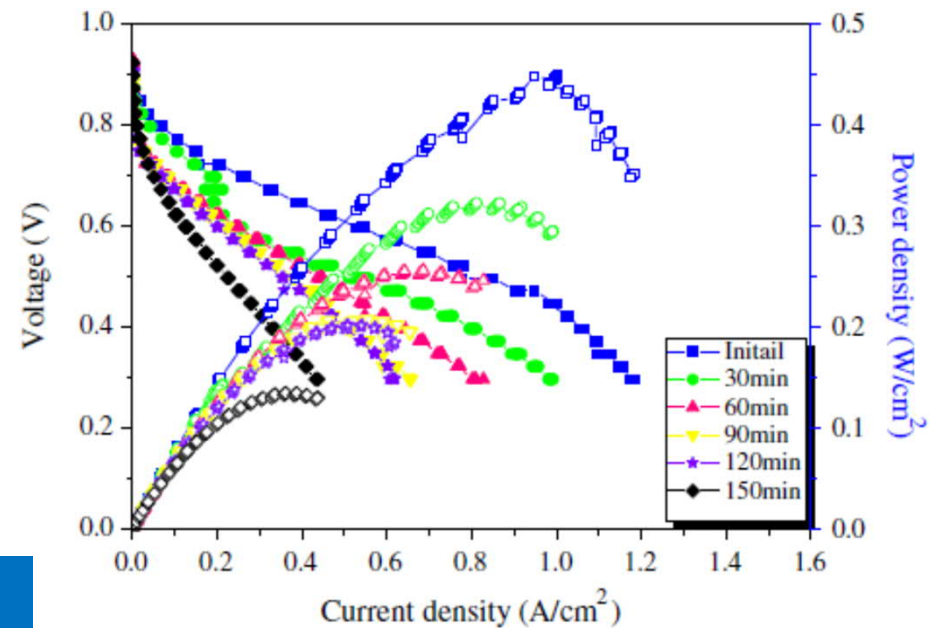
# How to measure durability?

## Accelerated durability test

- Speed up the degradation
- Unrealistic conditions?

Case: PEM FuelCell

At start-stop cycles a cell can  
Jump to 1.5 V (for seconds)



### Lecture Journal

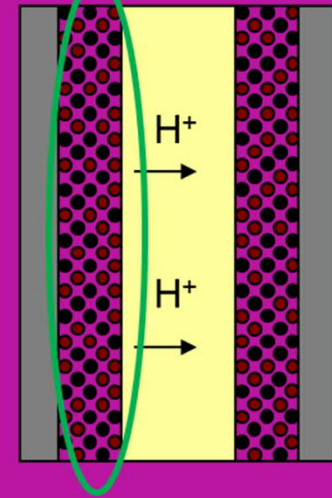
What do you think, are these tests useful? What else could we do?

G.-B. Jung et al. Applied Energy 100 (2012) 81-86.



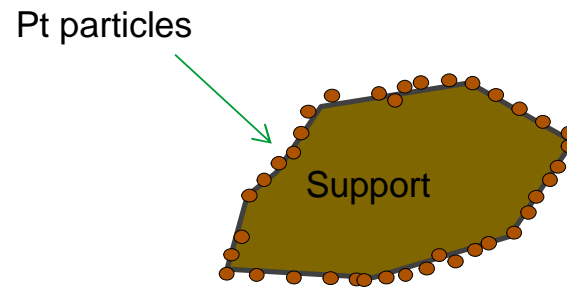
# Degradation mechanisms of nanoparticles

Active material: Catalyst



# Durability – High activity catalyst

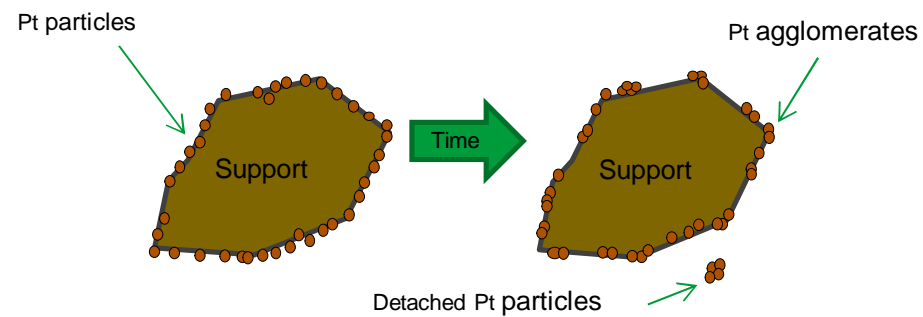
- High specific surface area
  - A lot of surface sites for the reaction to occur



- Need for support material
  - very porous
  - high surface area
  - conductive (carbon)

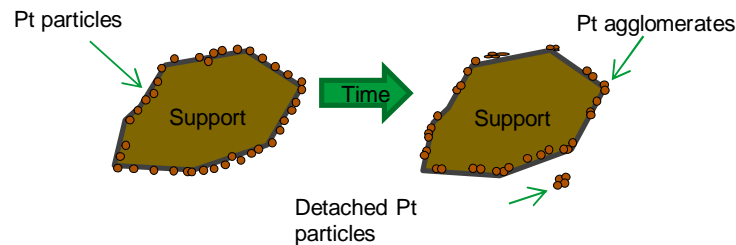
# Durability – High activity catalyst

- High specific surface area
  - >strong driving force to form larger agglomerants



# Durability – Nanoparticles

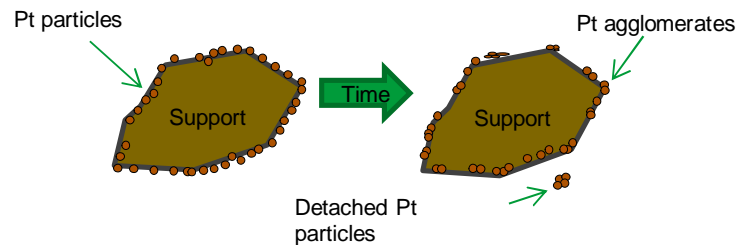
- High specific surface area
  - >strong driving force to form larger agglomerants



- Three mechanisms for agglomeration:
  - Ostwald ripening
    - Small particles dissolve in the ionomer phase and redeposit on surface of larger particles

# Durability – Nanoparticles

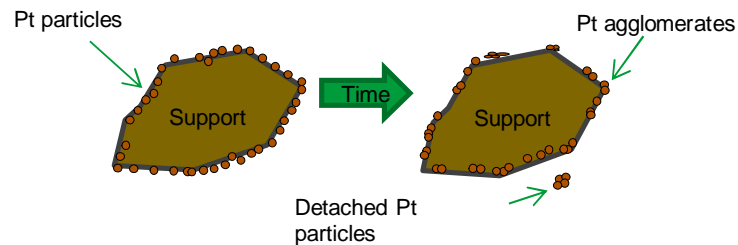
- High specific surface area
  - >strong driving force to form larger agglomerants



- Three mechanisms for agglomeration:
  - Ostwald ripening
    - Small particles dissolve in the ionomer phase and redeposit on surface of larger particles

# Durability – Nanoparticles

- High specific surface area
  - >strong driving force to form larger agglomerants



- Three mechanisms for agglomeration:
  - Ostwald ripening
  - Cluster-cluster collision
  - At the atomic scale minimizing clusters' Gibbs energy

# Nanoparticle Agglomeration

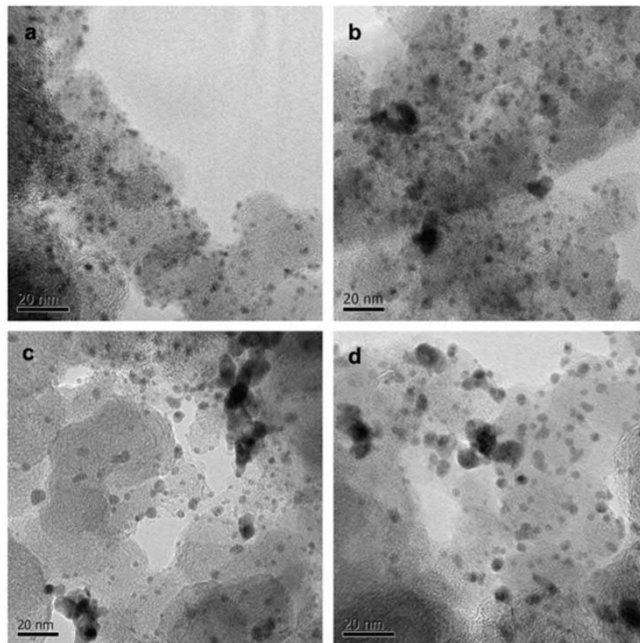


Fig. 6 – TEM images of the Pd/C at 0 h (a), 100 h (b), 300 h (c) and 500 h (d).

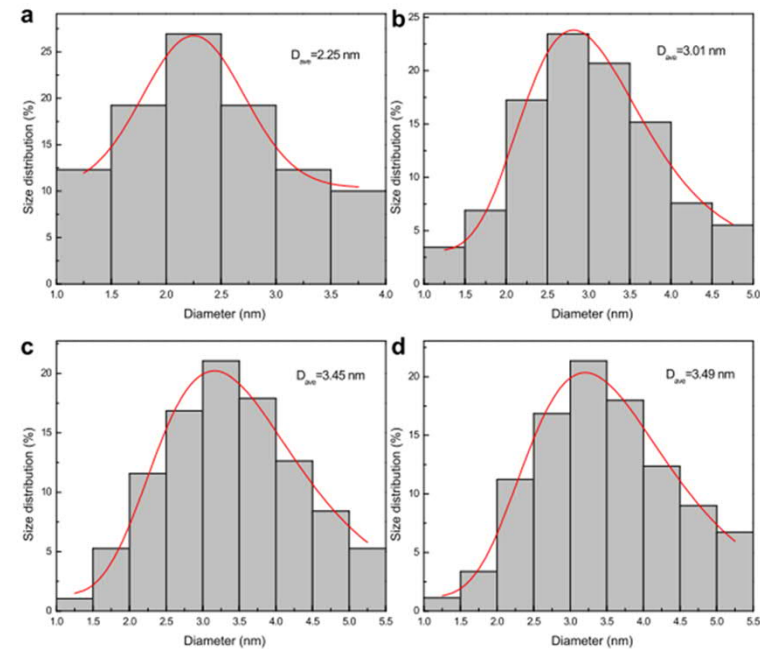
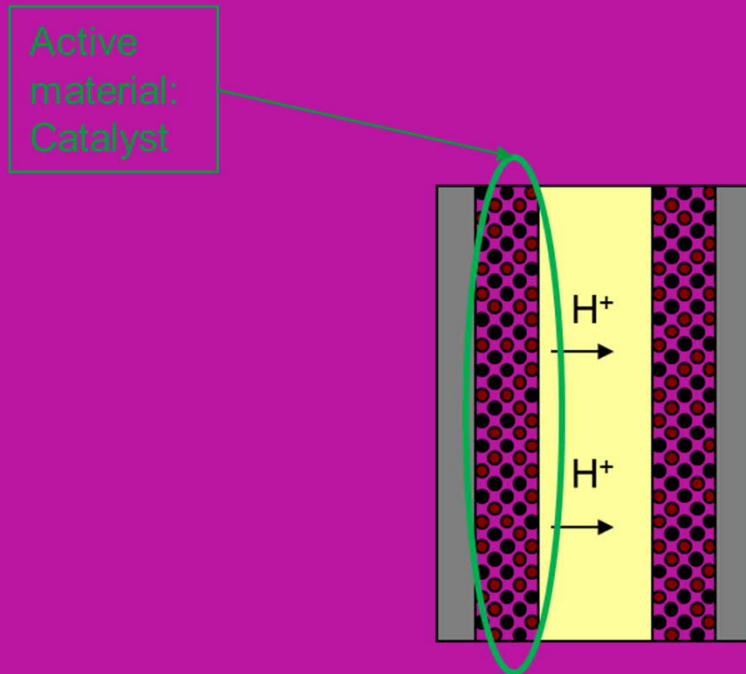


Fig. 7 – Histograms of Pd particle size distribution at 0 h (a), 100 h (b), 300 h (c) and 500 h (d).

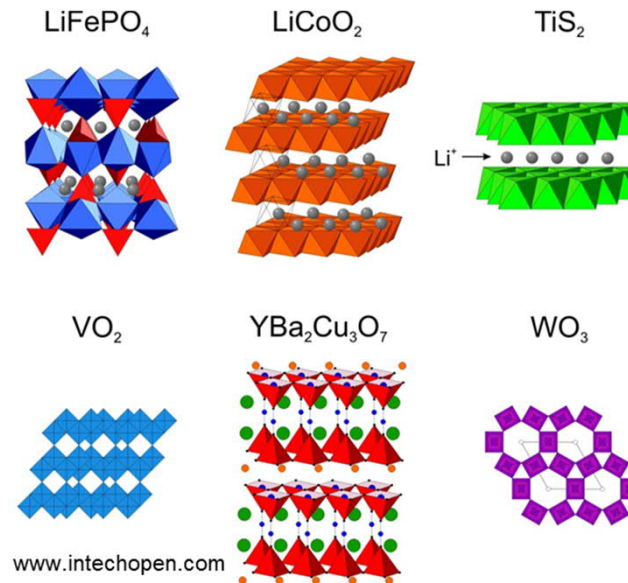
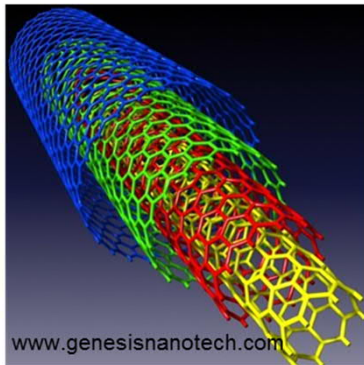
# Degradation challenges on 3D materials





# 3D Materials

- Structures
  - Carbon nanotubes
  - Li-ion materials
  - Do they collapse?



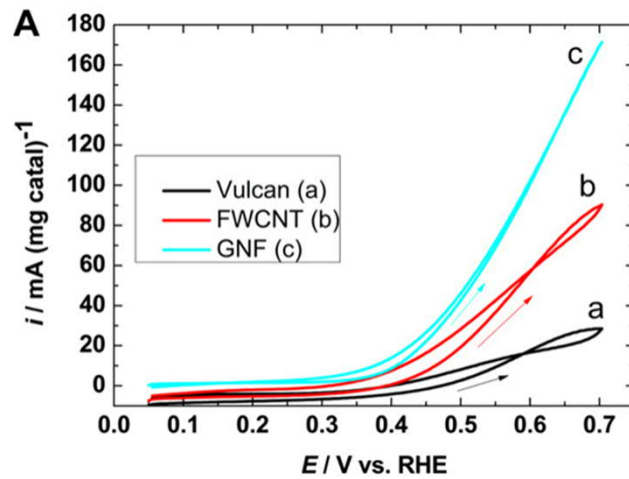
Positive electrode materials for  
Li-ion batteries

## Lecture Journal

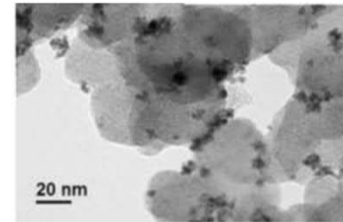
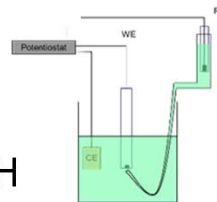
Reflect, how do we know/ensure  
the 3D material durability?

# 3D: Materials

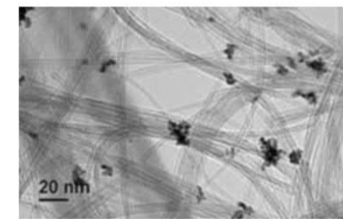
## Durability and Activity



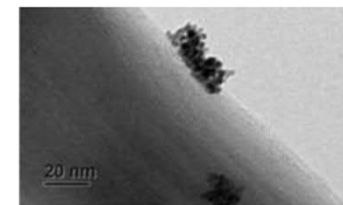
In electrochemical cell:  
 25 °C  
 0.1 M HClO<sub>4</sub> + 1 M MeOH



PtRu/Vulcan



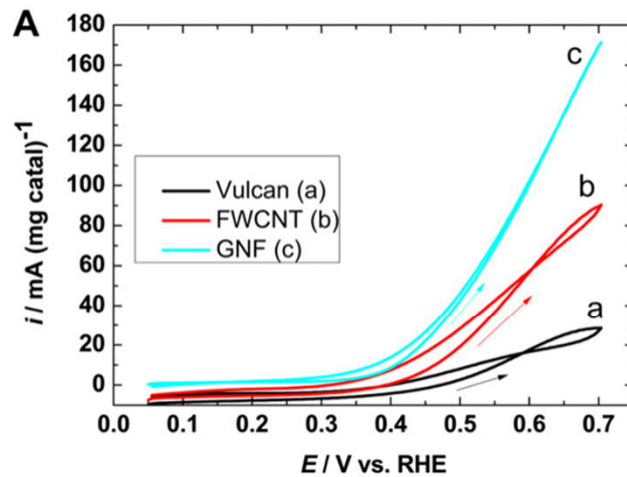
PtRu/FWCNT



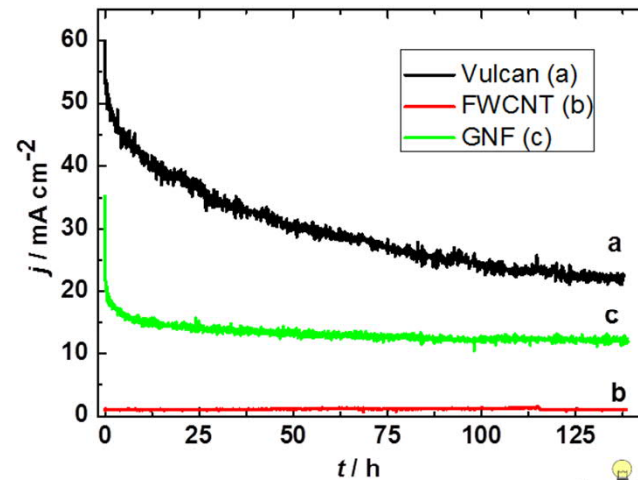
PtRu/GNF

# 3D: Materials

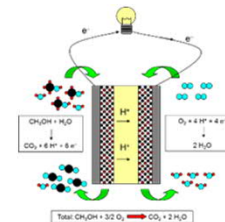
## Durability and Activity



In electrochemical cell:  
 25 °C  
 0.1 M HClO<sub>4</sub> + 1 M MeOH



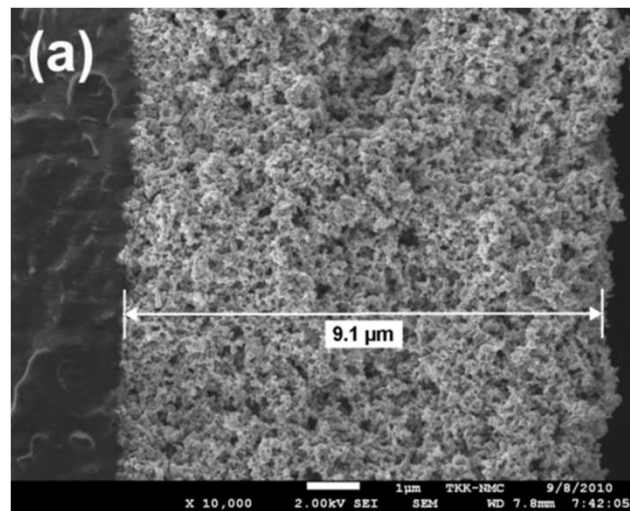
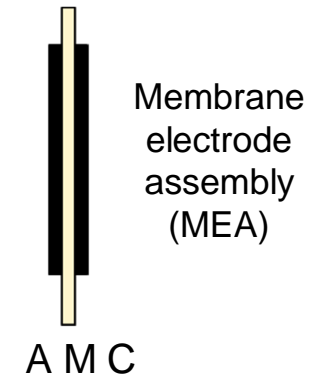
Single cell fuel cell:  
 70 °C  
 1 M MeOH



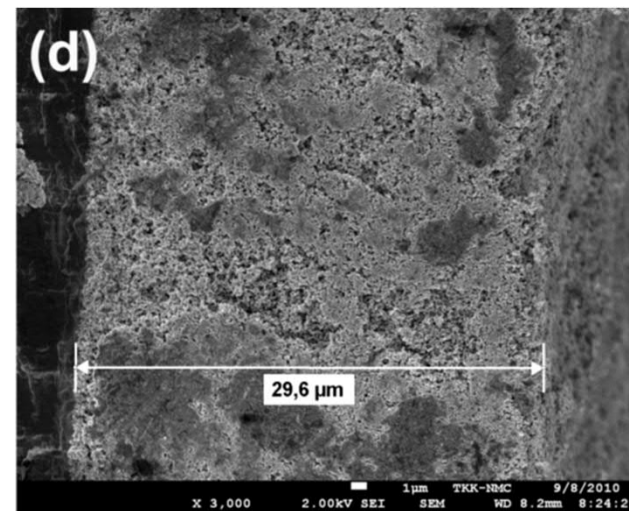
# 3D: Materials

## Durability and Activity

- PtRu/Vulcan anode



New

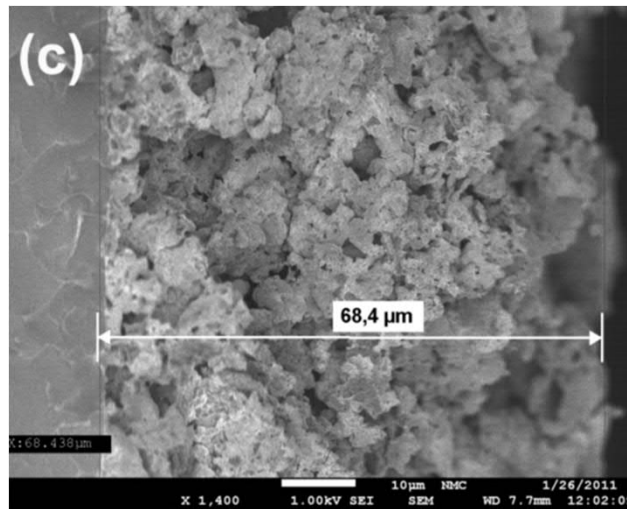


After 130 h

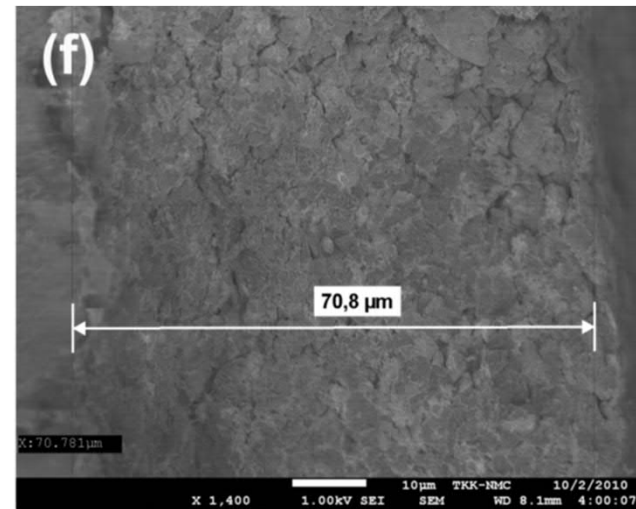
# 3D: Materials

## Durability and Activity

- PtRu/FWCNT anode



New

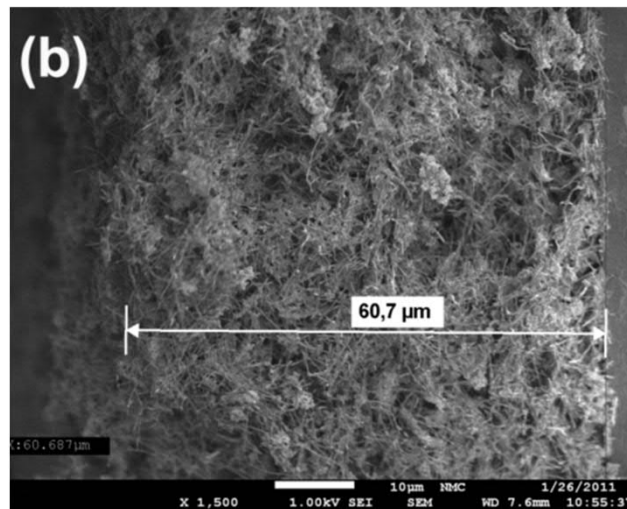


After 130 h

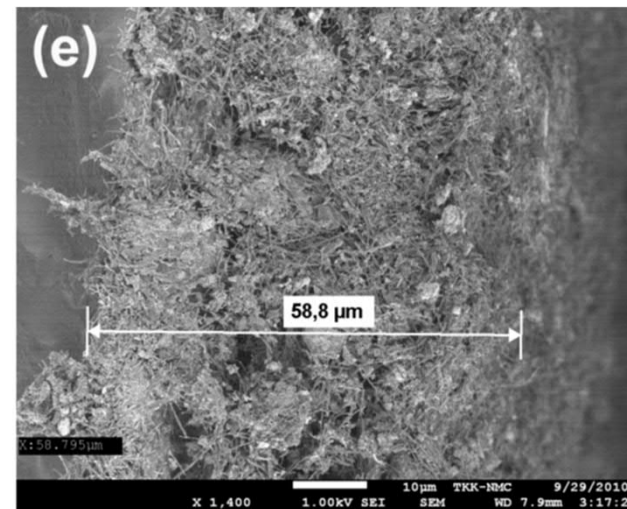
# 3D: Materials

## Durability and Activity

- PtRu/GNF anode



New

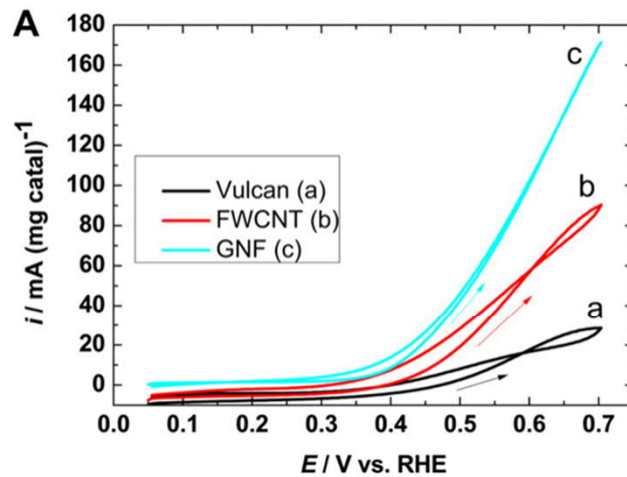


After 130 h

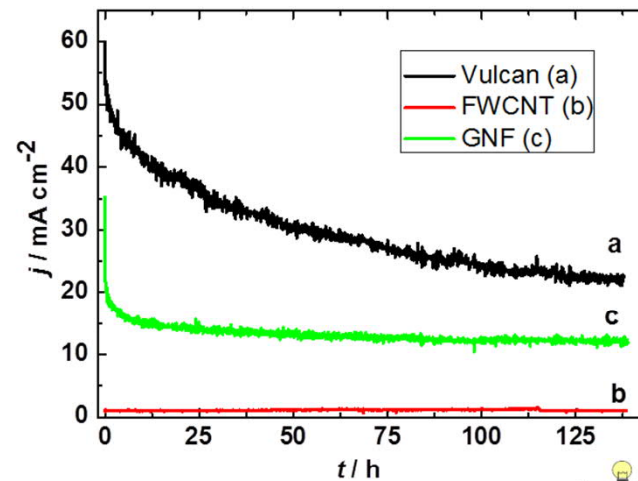


# 3D: Materials

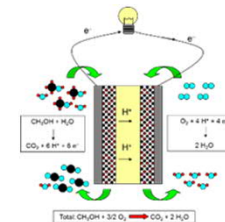
## Durability and Activity



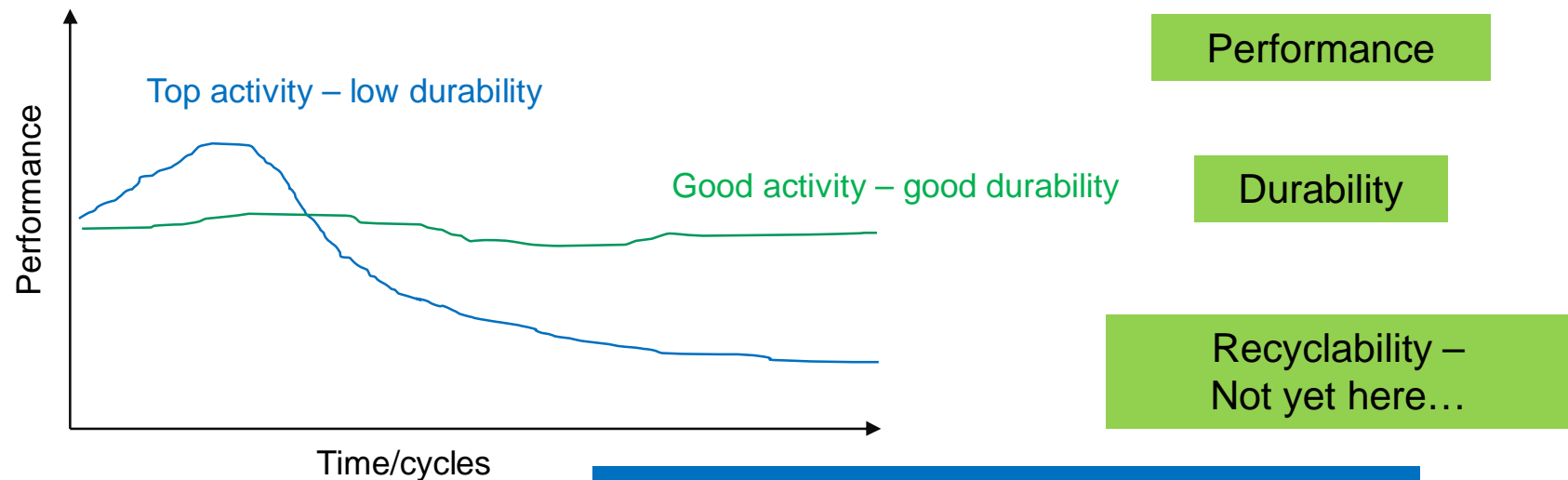
In electrochemical cell:  
 25 °C  
 0.1 M HClO<sub>4</sub> + 1 M MeOH



Single cell fuel cell:  
 70 °C  
 1 M MeOH



# What is valued in active material performance?



**Lecture Journal**

Reflect, as a consumer – which application would you choose?



## Take a home message

At science, there are large interest to produce exotic active materials for these solutions. But, *thermodynamics* is against that – sometimes, robust systems are most efficient (as they are stable).